

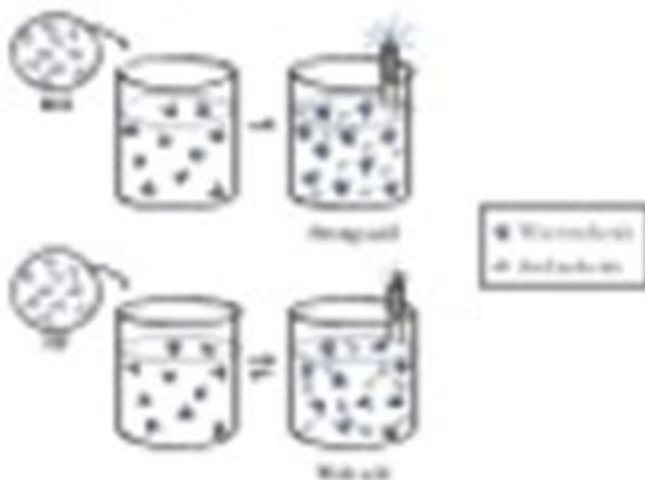
Strong versus Weak Acids

What makes an acid strong or weak?

What?

Strong acids are those that dissociate completely in water. The weak acids only partially dissociate. In other words, strong acids are those that dissociate completely in water. Weak acids are those that do not dissociate completely in water. The degree of dissociation is a measure of the strength of an acid.

Model - Acid Strength and Conductivity



What happens to the conductivity of a solution when you add a strong acid?

1. Explain the conductivity of the solution when you add a strong acid.

- _____ _____ _____
 - _____ _____ _____
- Weak electrolyte
 Strong electrolyte
 Strong acid
 Weak acid

Strong versus Weak Acids

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