

Worksheet Balancing Word Equations Chapter 10

(Remember the following are diatomic: H₂, N₂, O₂, F₂, Cl₂, Br₂, I₂)

The coefficients should add up to the number at the end that is in parenthesis.

1. carbon + oxygen → carbon dioxide (0)
2. copper + silver nitrate → copper(II) nitrate + silver (4)
3. zinc + copper(II) sulfate → zinc sulfate + copper (0)
4. mercury(II) nitrate + ammonium sulfide → mercury(II) sulfide + ammonium nitrate (2)
5. iron(III) hydroxide → iron(III) oxide + water (5)
6. water → hydrogen + oxygen (4)
7. copper(II) nitrate + water → hydrogen nitrate + copper(II) hydroxide (4)
8. phosphorus + oxygen → diphosphorous pentoxide (11)
9. iron + hydrochloric acid → iron(II) chloride + hydrogen (2)
10. sodium hydroxide + phosphoric acid → sodium phosphate + water (6)
11. antimony + chlorine → antimony trichloride (7)
12. sodium chlorate → sodium chloride + oxygen (7)
13. sodium carbonate + water → sodium oxide + hydrogen carbonate (0)
14. silver + sulfur → silver sulfide (2)
15. potassium + water → potassium hydroxide + hydrogen (6)
16. barium chloride + magnesium sulfate → barium sulfate + magnesium chloride (0)
17. sodium bromide + chlorine → sodium chloride + bromine (4)
18. ammonium carbonate → ammonia (NH₃) + carbonic acid(2)
19. calcium carbonate + hydrochloric acid → calcium chloride + hydrogen carbonate (2)
20. zinc chloride + phosphoric acid → zinc phosphate + hydrogen chloride (11)
21. mercury(II) oxide → mercury + oxygen(4)
22. sodium sulfite + acetic acid → hydrogen sulfite + sodium acetate (4)
23. ammonium nitrate + potassium hydroxide → ammonium hydroxide + potassium nitrate (0)
24. silver sulfate + aluminum chloride → silver chloride + aluminum sulfate (11)
25. barium sulfide + water → barium oxide + hydrogen sulfide (0)

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Balancing Word Equations Worksheet Answers

If you also get perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

49 Balancing Chemical Equations Worksheets [with Answers]

Worksheet Balancing Word Equations Chapter 10 (Remember the following are diatomic: H₂, N₂, O₂, F₂, Cl₂, Br₂, I₂) The coefficients should add up to the number at the end that is in parenthesis. 1. carbon + oxygen carbon dioxide (0) C + O₂ → CO₂ 2. copper + silver nitrate copper(II) nitrate + silver (4) Cu + 2 AgNO₃ → Cu(NO₃)₂ + 2 Ag 3. zinc + copper(II) sulfate zinc sulfate + copper (0)

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equation because chemical reactions must obey the Law of ____ of Matter. The number of atoms of each element on both sides of the equation must be ____ because matter cannot be ____ or _____. When balancing equations, the only numbers that can be changed are _____;

Worksheet: Balancing Equations Name I. Fill in the blanks ...

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Balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow. This is a collection of printable worksheets to practice balancing equations. The printable worksheets are provided in pdf format with separate answer keys.

How to Balance Equations - Printable Worksheets

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Worksheet #1: Writing and Balancing Formula Equations

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Chemical Reaction Equations Introduction Worksheet ...

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