

Check It

Without referring to the textbook, try to answer these questions. If you cannot answer a question, look up the relevant section in your textbook.

13.1 Oxidation Reactions

- Which conversion is an example of oxidation?
 - Carbon dioxide to carbon monoxide.
 - Concentrated hydrochloric acid to chlorine.
 - Copper(II) oxide to copper.
 - Sodium hydroxide to sodium nitrate. (B)
- Which gaseous substance has nitrogen in its **lowest** oxidation state?
 - N_2O
 - NH_3
 - NO
 - NO_2 (B)

13.2 Reduction Reactions

- Which underlined substance has been reduced?
 - N_2 to NH_3
 - NH_3 to NO
 - NH_3 to NO_3^-
 - NO_2 to NO_3^- (A)

13.3 Redox Reactions

- What happens to a bromine atom when it is changed to a bromide ion?
 - It is oxidised.
 - It is reduced. (B)
 - It loses a proton.
 - It loses an electron.
- Which reaction is **not** a redox reaction?
 - $2Fe^{3+}(aq) + 2I^{-}(aq) \rightarrow 2Fe^{2+}(aq) + I_2(aq)$
 - $2Na(s) + Cl_2(g) \rightarrow 2NaCl(s)$
 - $Ag^{+}(aq) + Cl^{-}(aq) \rightarrow AgCl(s)$ (C)
 - $Mg(s) + 2Ag^{+}(aq) \rightarrow Mg^{2+}(aq) + 2Ag(s)$

13.4 Oxidising and Reducing Agents

- Which substance cannot reduce iron(III) oxide to iron?
 - Carbon dioxide
 - Carbon monoxide
 - Coke (carbon)
 - Hydrogen (A)
- Why does the colour of potassium iodide change to brown when chlorine is bubbled through it?
 - A compound is formed between chlorine and potassium iodide.
 - Chlorine oxidises iodide ions to iodine. (B)
 - Iodine reacts with starch to give a blue colour.
 - Potassium chloride is formed which is coloured.

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Floyd Vogt is a sixth-generation carpenter and contractor. His family operated a small business devoted to all phases of home construction, and he began working in the family business at age 15. After completing a B.A. in chemistry at the State University of New York College at Oneonta, Mr. Vogt returned to the field as a self-employed remodeler. In 1985 he began teaching in the carpentry ...

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