

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

**CHAPTER 10 REVIEW***States of Matter***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. Identify whether the descriptions below describe an ideal gas or a real gas.
- \_\_\_\_\_ **ideal gas** a. The gas will not condense because the molecules do not attract each other.
- \_\_\_\_\_ **ideal gas** b. Collisions between molecules are perfectly elastic.
- \_\_\_\_\_ **real gas** c. Gas particles passing close to one another exert an attraction on each other.
2. The formula for kinetic energy is  $KE = \frac{1}{2}mv^2$ .
- a. As long as temperature is constant, what happens to the kinetic energy of the colliding particles during an elastic collision?  
**The energy is transferred between them.**
- b. If two gases have the same temperature and share the same energy but have different molecular masses, which molecules will have the greater speed?  
**Those with the lower molecule mass.**
3. Use the kinetic-molecular theory to explain each of the following phenomena:
- a. A strong-smelling gas released from a container in the middle of a room is soon detected in all areas of that room.  
**Gas molecules are in constant, rapid, random motion.**
- b. As a gas is heated, its rate of effusion through a small hole increases if all other factors remain constant.  
**As a gas is heated, each molecule's speed increases; therefore, the molecules pass through the small hole more frequently.**
4. a. \_\_\_\_\_ **b, d, c, a** List the following gases in order of rate of effusion, from lowest to highest. (Assume all gases are at the same temperature and pressure.)  
(a) He (b) Xe (c) HCl (d) Cl<sub>2</sub>

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Application 10.1.3.2 Dernière mise à jour le 11.10.2004 1/3 Chapitre 10 La fissuration du béton Section 10.1 Les principes de comportement 10.1.3.2 un exercice d'application Thème de l'exercice : étapes de bétonnage et joints Soit une dalle de bâtiment de 20 x 36 m supportée par des colonnes. La stabilité horizontale est assurée par deux noyaux en béton. • Fixer le concept des ...

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Dans un document, j'ai 3 sections avec des titres dont la numérotation devrait être indépendante: S1. avec une introduction (heading 1 sans numérotation sauf pour les subdivisions) S2. avec dix chapitres (heading 1 numérotés "Chapitre n" de 1 à 10) et les subdivisions numérotées hiérarchiquement dans le chapitre (1.1, 1.2, 1.2.1 etc)

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### Chapter 10 Review States Of Matter Section 1 Answer Key

10.1 Review. 10.11 Where are we? Numbers are numbers. Read the section on them again. Functions are sets of (argument, value) pairs of numbers. They are often described by formulae which tell us how to compute the value from the argument. Only one value is allowed for each argument. The formulae you will usually encounter start with the ...

### 10.1 Review - math.mit.edu

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section and a second column for his contribution to evolutionary theory. Scientist Contribution Linnaeus Buffon 10.1 Early Ideas About Evolution KEY CONCEPT There were theories of biological and geologic change before Darwin. MAIN IDEAS • Early scientists proposed ideas about evolution. • Theories of geologic change set the stage for Darwin's theory. VOCABULARY evolution, p. 298 species ...

### CHAPTER 10 Principles of Evolution

Section 10 - Obligations des dirigeants, des personnes concernées par l'offre et de leurs conseils. Règlement général en vigueur du 01/09/2012 au 30/09/2012. Livre I-L'Autorité des marchés financiers. Titre I-Fonctionnement de l'Autorité des marchés... Chapitre 1er-Déontologie des membres de... Chapitre 2-Déontologie des experts nommés auprès... Chapitre 3-Indemnités et ...

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