

YOU'D BREAK THE PH SCALE

14

Alkali metals

metals

1) $\text{HCl(aq)} + \text{Na}_2\text{O} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$

2) $\text{SO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_3(\text{aq})$

3) $\text{MgO} + \text{H}_2\text{O} \rightarrow \text{Mg(OH)}_2(\text{aq})$

4) $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3(\text{aq})$

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